CHAPTER 1 Introduction to Matter CHAPTER | INTRODUCTION TO MATTER

Vocabulary

| matter | molecule         | solid  |                 |
|--------|------------------|--------|-----------------|
| mass   | element          | liquid | 1               |
| weight | compound         | gas    |                 |
| volume | mixture          |        |                 |
| atom   | states of matter |        | 5 <b>%</b><br>5 |

A. WHO AM I?

Write the vocabulary word on the line that best matches each clue.

- **1.** I'm a substance that has a fixed volume but not a fixed shape.
- 2. I'm a substance that results when two or more different types of atoms bond together.
- **3.** I'm a particle made up of two or more atoms bonded together.
- 4. I'm a substance that has no fixed volume or shape.
- **5.** I'm a combination of different substances that remain the same individual substances and can be separated by physical means.
- 6. I'm a substance that has a fixed volume and a fixed shape.
- 7. I'm a substance that contains only a single type of atom.
- 8. I'm the measure of how much matter an object contains.
- 9. I'm the smallest basic unit of matter.

- 11. I'm the downward pull of an object due to gravity.
- 12. I'm anything that has mass and takes up space.
- **13.** I'm the amount of space that matter in an object occupies.

## **B. WHICH ONE?**

Choose the vocabulary word that matches each description and write it on the line. Use the bolded words in the sentences as clues.

14. This can be made of atoms that are different or atoms that are alike.

molecule

element

state of matter

15. This does not always contain the same percentages of the various substances.

Element

compound

mixture

**16.** The particles in this are not close to one another and can move about easily in any direction.

molecule

solid

gas

17. The number of atoms in this in not important as long as all of the atoms are of the same type.

matter

mixture

element

18. This takes on the shape of the container it is in.

atom

liquid

solid